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## **3 Atoms: The Building Blocks of Matter**

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1. c When an electron is added to a neutral atom, energy is (a) always absorbed. (c) either absorbed or released. (b) always released. (d) neither absorbed nor released.
2. d The energy required to remove an electron from a neutral atom is the atom's

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CHAPTER 9 REVIEW Stoichiometry  
SECTION 3 PROBLEMS Write the answer  
on the line to the left. Show all your work  
in the space provided. 1. 88% The actual  
yield of a reaction is 22 g and the  
theoretical yield is 25 g. Calculate the  
percentage yield. 2. 6.0 mol of N<sub>2</sub> are

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mixed with 12.0 mol of H

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Chapter Review. p.59. Problem Bank.  
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CHAPTER 17 REVIEW Reaction Kinetics  
MIXED REVIEW SHORT ANSWER Answer



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the following questions in the space provided. 1. The reaction for the decomposition of hydrogen peroxide is  $2\text{H}_2\text{O}_2(\text{aq}) \rightarrow 2\text{H}_2\text{O}(\text{l}) + \text{O}_2(\text{g})$ . List three ways to speed up the rate of decomposition. For each one, briefly explain why it is effective, based on collision theory.

## **17 Reaction Kinetics - David Brearley High School**

164 Study Guide for An Introduction to Chemistry Section Goals and Introductions Section 11.1 The Mysterious Electron Goals To explain why it is very difficult to describe the modern view of the electron. To give you some understanding of the nature of the electron by describing how it is like a guitar string. To explain what atomic orbitals are.

## **Chapter 11 Modern Atomic Theory - An Introduction to Chemistry**

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